

1 lower section 20. The reaction chamber has outer
2 diameters ranging from $\frac{1}{4}$ inch to 4 inches.

3 An axially extending radiant burner 7 is
4 vertically disposed along the central axis of the
5 helical coil section 20 of the tubular reaction
6 conduit. The radiant burner is supported by a burner
7 gas conduit 12 that conveys a mixture of fuel and
8 oxidant from the inlet means 8 to the radiant burner.
9 In this embodiment, the radiant burner 7 comprises a
10 gas permeable metal fiber zone 14 that subtends the
11 entire circumference of the radiant burner. Fuel and
12 oxidant pass through the permeable metal fiber zone 14
13 where they are ignited on the surface, thereby
14 combusting and releasing heat to form an incandescent
15 zone that radiates energy in a predominantly uniform
16 radial direction. The helical tubular reaction chamber
17 and catalyst therein are sized for creation of mass
18 velocities ranging from 400 lb/ft²/h to 1500 lb/ft²/h.
19 The catalyst in the helical tubular reaction chamber
20 has average catalyst particle diameters ranging from $\frac{1}{4}$
21 to 1 inch for producing gas pressure drops ranging from
22 1 psi to 8 psi during flow through the reaction
23 chamber. The helical tubular reaction chamber has gas
24 exit end temperature ranging from 1150°F to 1400°F,
25 when heated by said radiant burner, in operation. The
26 helical tubular reaction chamber has maximum tube wall

1 temperatures ranging from 1300°F to 1600°F, when heated
2 by said radiant burner, in operation. The helical
3 tubular reaction chamber has average heat fluxes
4 ranging from 3,000 btu/ft²/h to 10,000 btu/ft²/h, when
5 heated by said radiant burner in operation. The
6 helical tubular reaction chamber is sized to have
7 capacity to generate hydrogen plus carbon monoxide
8 product in volumetric quantities ranging from 50 SCFH
9 to between 100 and 1500 SCFH. The radiant burner
10 comprises a supported metal fiber material consisting
11 essentially of an alloy containing principally iron,
12 chromium, and aluminum and smaller quantities of
13 yttrium, silicon, and manganese, said alloy having
14 extended life at operating temperatures up to 2000°F.
15 The radiant burner has surface temperatures ranging
16 between 1500°F and 1900°F, in operation. The radiant
17 burner has an operating combustion intensity typically
18 ranging from 150,000 btu/ft²/h to 350,000 btu/ft²/hr,
19 wherein the combustion intensity is defined as the
20 higher heating value of the fuel combusted divided by
21 the permeable radiant burner surface area. The radiant
22 burner has an operating excess air ratio typically
23 ranging from 30% to 100%, wherein the excess air ratio
24 is defined as percent combustion air in excess of the
25 stoichiometric amount required for complete combustion
26 of the burner fuel. The helical coil has free area in

1 the range 50% to 75%, wherein the free area is defined
2 as the ratio of the free area between successive coil
3 turns and the cylinder that bisects the helical coil
4 circle.

5 In Figs. 1, 3 and 4, a gas conditioning
6 system 101 and fuel cells 100 to receive hydrogen are
7 in operative communication with reactor outlets 3.

8 Fig. 5 depicts yet another embodiment of the
9 present invention. In this embodiment shown
10 schematically the reaction chamber 116 is defined by
11 the annular space between an outer conduit 131 and an
12 inner conduit 132. The reactant gases enter the
13 reaction chamber through inlet means 112, and pass
14 through catalyst bed at 116 and then to space 134 at
15 the inlet of the inner conduit 132. The reactant gases
16 exit the inner conduit space through exit means 113.
17 The reactant gases passing through the inner conduit
18 132 transfer heat to the reactant gases contained in
19 the reaction chamber 116 to beneficially recuperate
20 heat from the endothermic reaction.

21 An axially extending radiant burner 107 is
22 vertically disposed within a combustion chamber 104.
23 The radiant burner is oriented in parallel with the
24 longitudinal extent of the tubular reaction conduit.
25 If a multiplicity of such tubular reaction conduits are
26 used, they can be oriented concentrically around a